Evaluation of Minimum NO_x Emission from Ammonia Combustion

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Introduction

- Ammonia is a potential carbon-free fuel
- Increased NO_x concern with fuelbound nitrogen
- This work explores staged combustion for low NOx
- Rich-Quench-Lean (RQL) used in systems with high turndown (aviation) is promising for ammonia
 - Previous studies¹ observed emissions as low as 50 ppm NO_x emissions
 - Previous studies¹ showed large amounts of H₂ produced in rich ammonia flames (over 3000 ppm)



Innocenti, A., 2016. Numerical analysis of the dynamic response of practical gaseous and liquid fuelled flames for heavy-duty and aero-engine gas turbines.

¹R.C. Rocha, M. Costa, X.-S. Bai, Combustion and Emission Characteristics of Ammonia under Conditions Relevant to Modern Gas Turbines, Combustion Science and Technology 193 (2021) 2514-2533.



Motivation

- Research Question: what is the theoretical minimum possible NO_x emission from ammonia combustion?
 - Not simulating a specific combustor
 - Addressing what is possible with technological development
 - "NOx entitlement" for ammonia RQL
- Minimization problem applied to RQL architecture
 - Vary combustor parameters
 - Minimize NO_x low while limiting H_2 emissions
- Analyze sensitivities to firing temperature, combustor pressure, global residence time, and RQL parameters



NOx entitlement for lean premixed natural gas.

Reproduced from Leonard, G. and Stegmaier, J., 1994. Development of an aeroderivative gas turbine dry low emissions combustion system.

Approach

- Revisiting previous staged combustor NOx entitlement modeling of Goh et al.
- Modeled using reactor network model in Cantera
- Each stage represented as 1-D flame with perfect mixing assumption
- Perfect mixer is an adiabatic, nonreacting constant pressure batch reactor

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Prediction of minimum achievable NO_x levels for fuel-staged combustors



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Reactor Network Model

- 1-D flame calculates output for a specified residence time
 - Model calculates in spatial coordinates, conversion to temporal coordinates done after
 - NH_2 peak used to define τ_0 for each stage
- Constants:
 - Fuel: NH₃
 - Oxidizer: 79% N₂, 21% O₂
- Variables:
 - Texit (determined by Φ_{global})
 - Combustor pressure
 - Total residence time, τ_{global}



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Reaction Mechanism

- Compared multiple kinetic models capable of ammonia combustion and NO_x formation
- Chosen mechanism: Mei et al. (2019)
 - Best agreement with experimental datasets
 - Includes all NO_x formation and NH_3 oxidation routes



Optimization Problem and Constraints

- Parameters: Φ_{main} and τ_{sec}
 - Find optimal parameters that result in minimum NO
- H₂ is a significant product of rich main stage
 - Present at τ_0 since NH₂ peak is used to define start of flame
- Optimizer finds lowest possible τ_{sec} to meet H₂ emissions constraint (combustion efficiency)



Evolution of $NH_3 \& H_2$ through rich ammonia-air flame ($\Phi = 1.2, P = 1$ bar)

NO Dependence on Φ_{main} and τ_{sec}

- NO emissions sensitive to staging parameters even for fixed $\Phi_{global}~$ and τ_{global}
- Staging parameters are

 $- \ \Phi_{\text{main}}$

- $au_{secondary}$

- Example: NOx vs residence time for two secondary zone residence times
- NOx relaxes towards equilibrium in rich zone
- NOx rises in lean secondary zone
 - Shorter lean zone produces less NOx
 - Lean zone necessary for H₂ burnout (combustion efficiency)





- Example: NO and H2 emissions at fixed Φ_{global} , τ_{global} , and τ_{sec}
- There is an optimum, rich main stage equivalence ratio
- H₂ emission is nearly constant, indicating oxidation can be completed in less than 5 ms for this condition



$\tau_{\text{sec}} \, \text{Dependence}$

- NO and H₂ have an inverse relationship
 - Long second stage oxidizes H₂
 - Long second stage produces NO

- Objective: shorten τ_{sec} as much as possible within H₂ constraint
 - Example for T_{exit} = 2050 K
 - In this example, $\tau_{sec} = 3$ ms is optimal



NO Production Sources

- Unrelaxed main stage NO can be a significant NO_x contribution
- Residence time required for equilibrium much larger than current practical combustor designs
 - 1000 ms at 1 bar
 - 100 ms at 10 bar
 - Pressure helps
- Two step relaxation due to OH formed in rich main stage reacting with N and NH to form NO





NO Production Sources

- High pressure and richer main stage will drive NO_{eq,main} down
- Observation: achieving equilibrium NOx in rich main stage enables very low NOx
 - Lean secondary zone NOx must be limited
 - What is the main source of NO emissions in ammonia RQL?



Combustor Firing Temperature Sensitivities

- Majority of NO contribution is from NO_{main,unrelaxed}
- Firing temperature helps: faster main stage NO relaxation
- Longer main stage would help



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Combustor Firing Temperature Sensitivities

- Richer main stage beneficial (lower equilibrium NO)
- Relaxation toward
 equilibrium faster with hotter
 main stage
- Higher temperatures allow for faster H₂ oxidation
 - Shorter τ_{sec}
 - Helps with Zeldovich NOx



Optimized parameters for minimum NO at various firing temperatur



Combustor Pressure Sensitivities

- High pressure allows main stage to approach equilibrium faster
- Firing temperature more sensitive at lower pressures



Combustor Pressure Sensitivities

• Optimal Φ_{main} and τ_{sec} are independent of firing temperature at high pressure



Optimal parameters for minimum NO at varying combustor pressure



Global Residence Time Sensitivities

- Longer combustor residence time helps
- Main stage NO can relax towards equilibrium



Global Residence Time Sensitivities

- Optimum Φ_{main} same for all conditions (high pressure)
- Optimal τ_{sec} independent of global residence time
 - All added residence time is main stage, reducing NO_{main,unrelaxed}



Conclusions

- Theoretical minimum NOx emissions below 30 ppm are possible for ammonia combustion with RQL
- Intuition for NOx formation from lean premixed combustion is not applicable here
- Low NOx rich main stage (relax to equilibrium):
 - High temperature
 - High pressure
 - Long residence time
- Low NOx secondary stage:
 - Low residence time
 - Sufficient residence time to burn out H₂ for combustion efficiency
- Significant engineering required to navigate real world effects and approach these limits



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